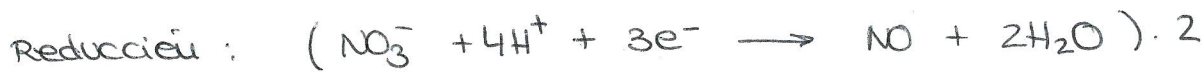
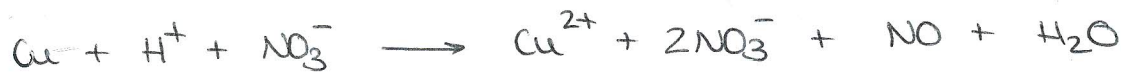
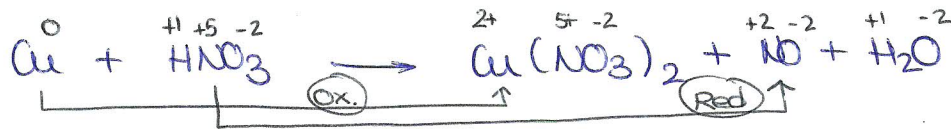
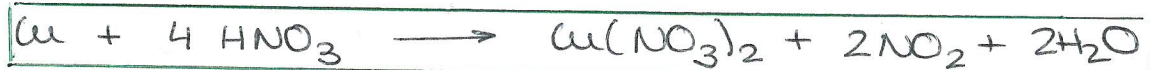
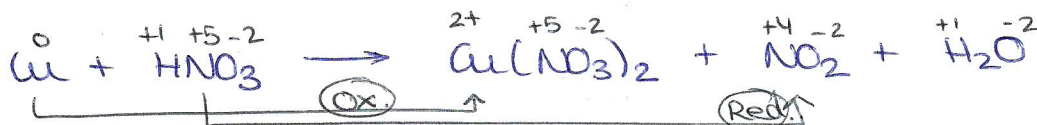


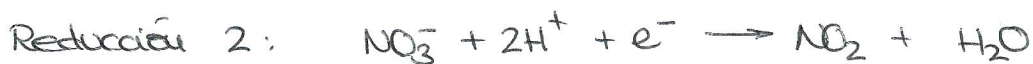
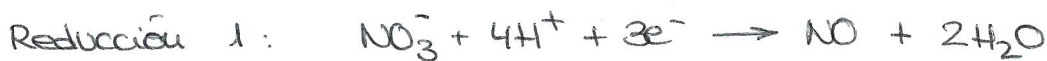
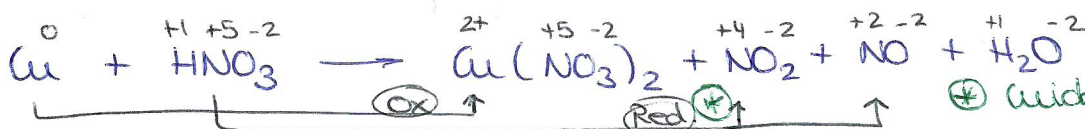
①



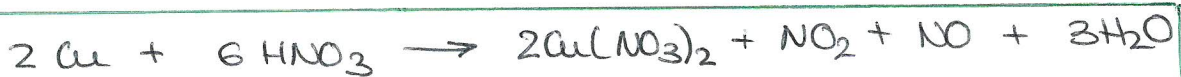
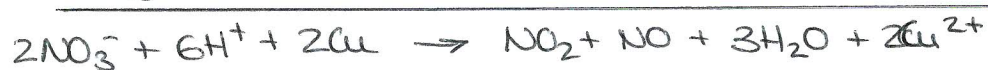
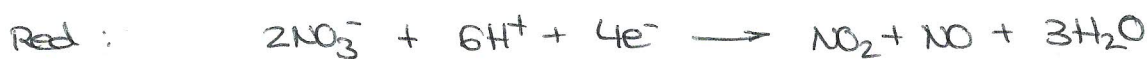
②



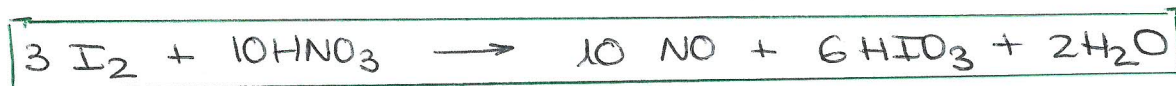
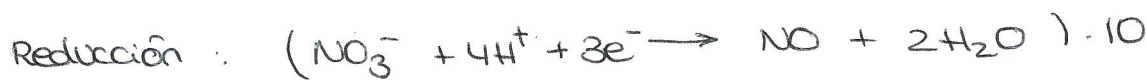
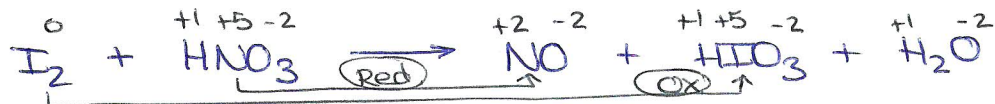
③



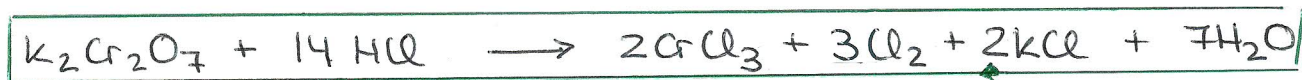
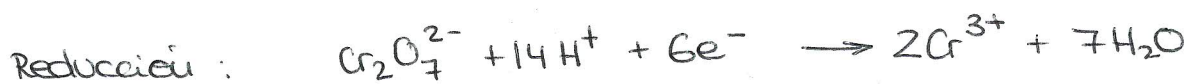
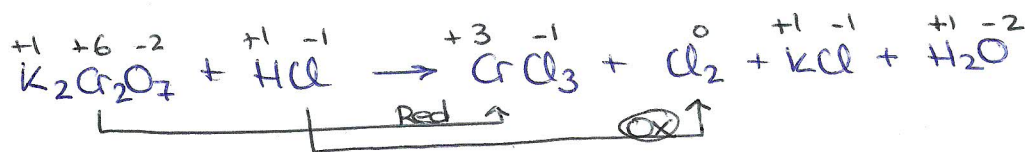
Sumamos Red. 1 + Red. 2 para tener una única semireacción de reducción:



④

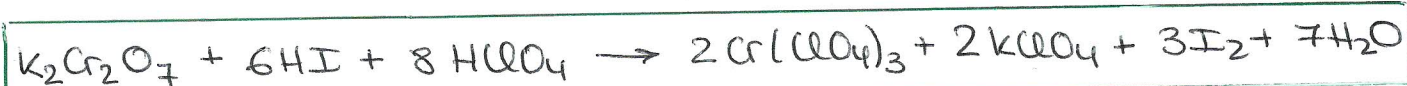
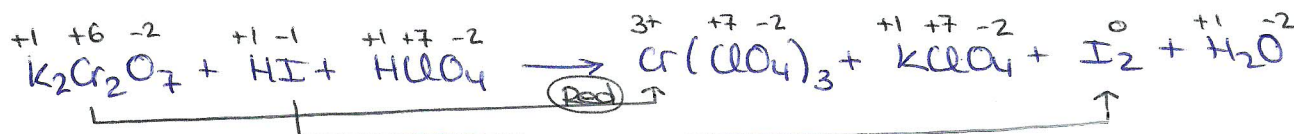


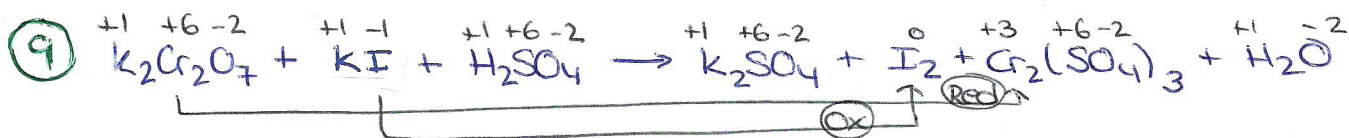
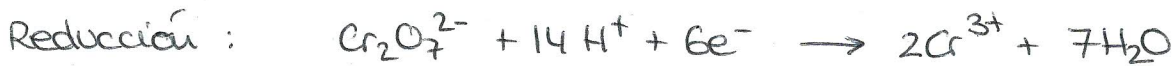
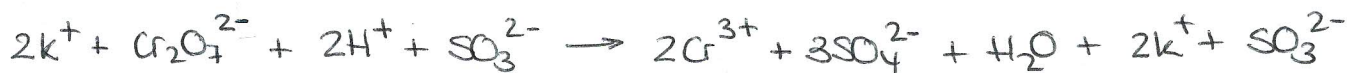
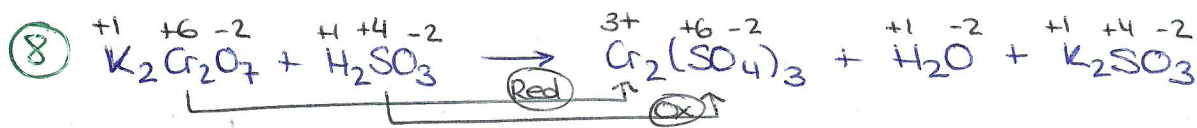
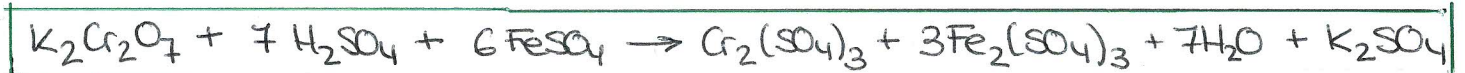
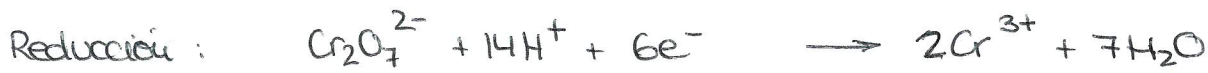
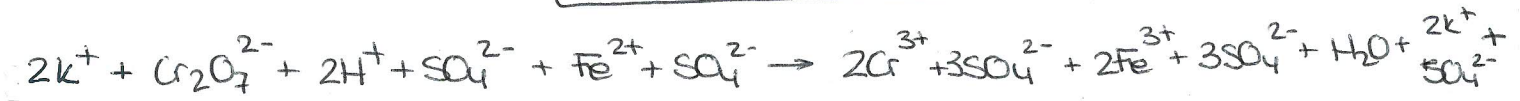
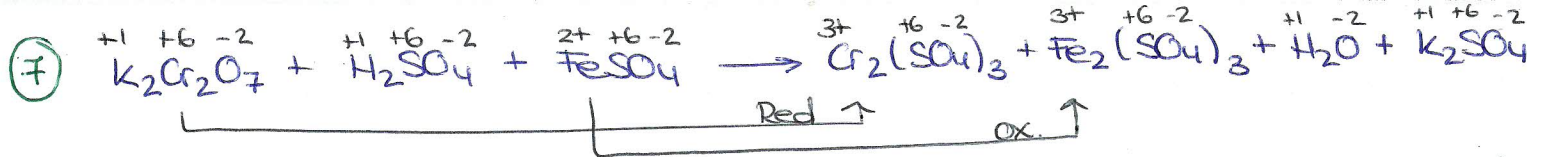
⑤

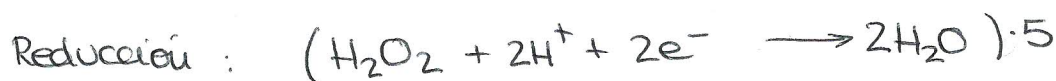
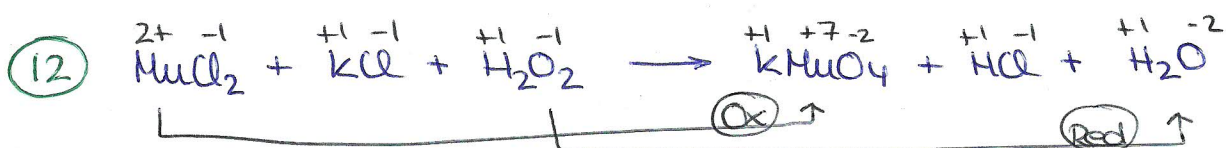
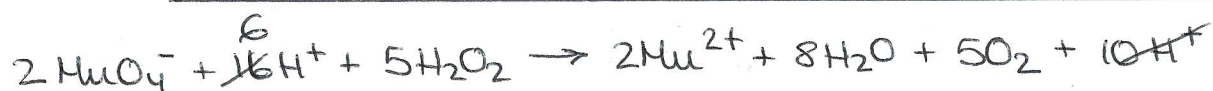
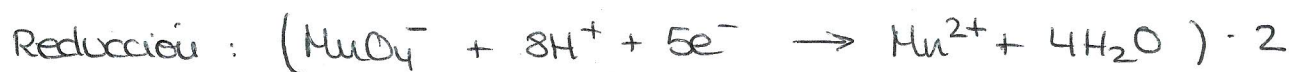
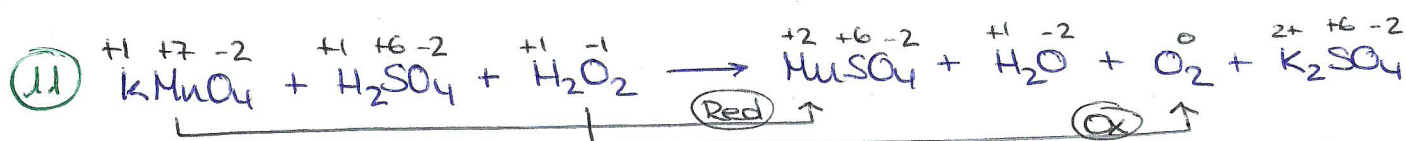
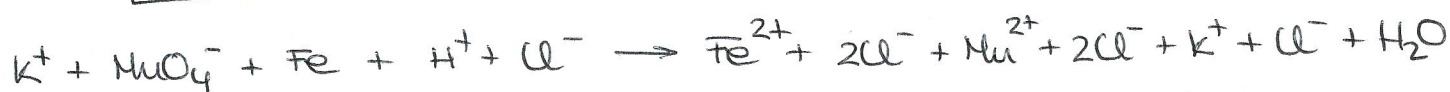
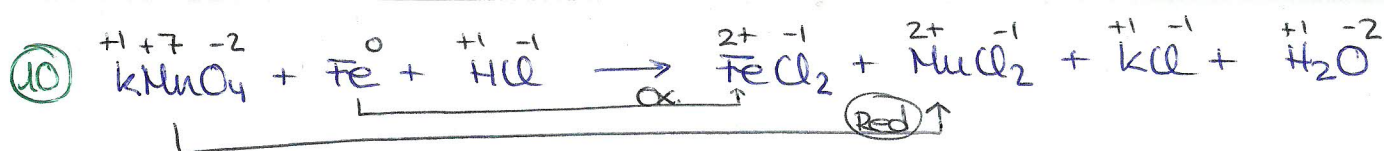


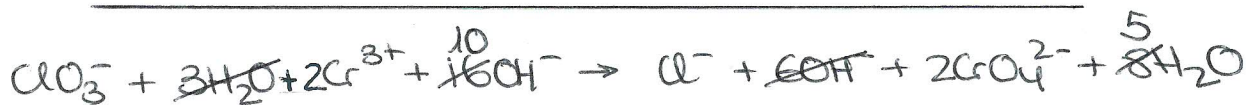
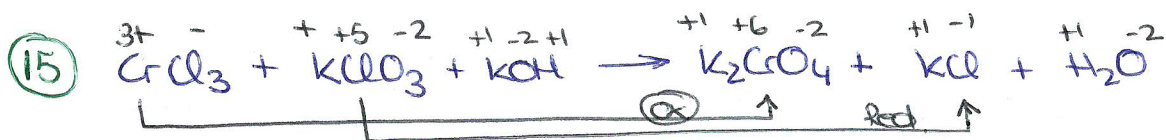
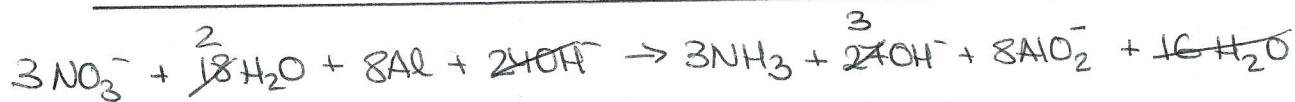
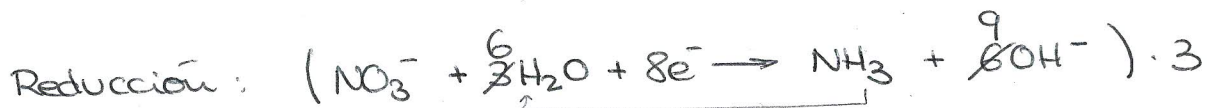
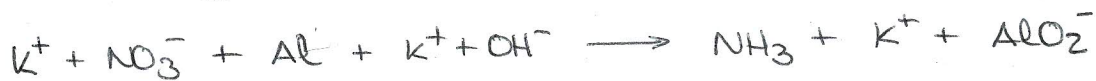
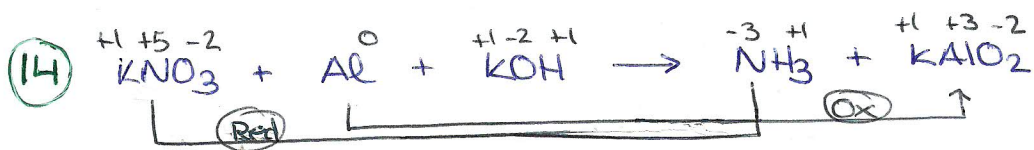
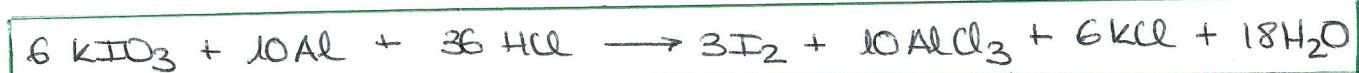
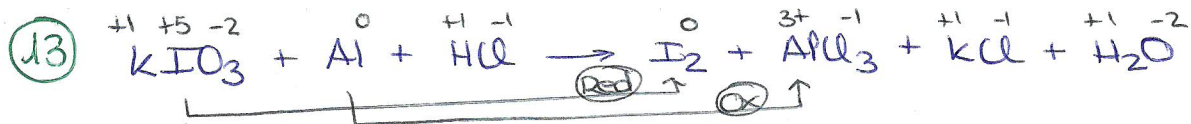
Este "2" se ajusta por el "K".

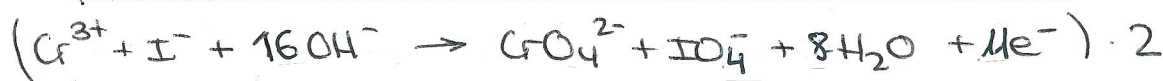
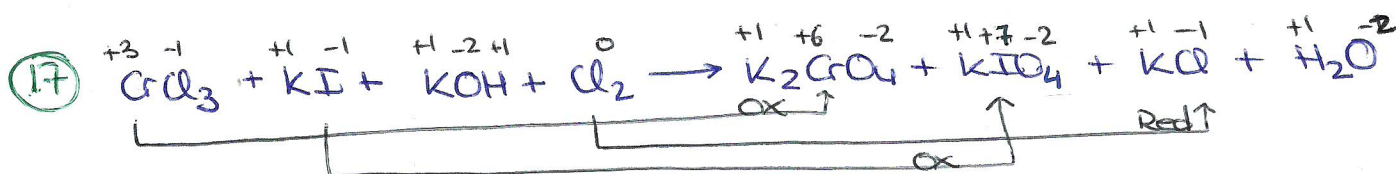
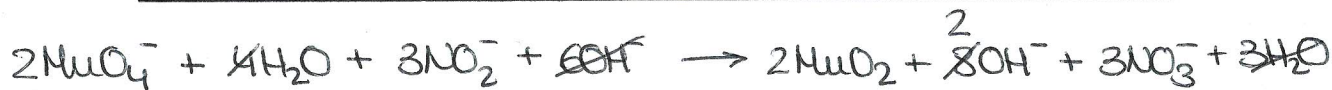
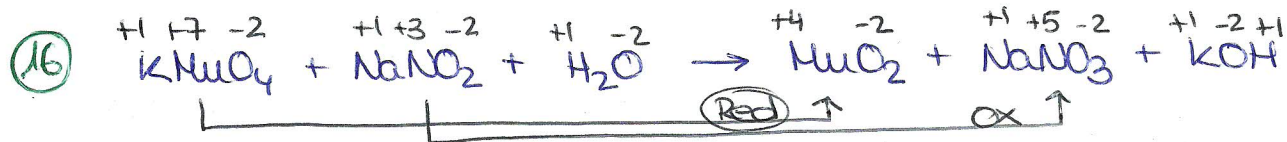
⑥











↑
los 22 Cl⁻ que salen
+ los 6 Cl⁻ de reactivos
que no participan en
la redox.